

L 19486-63 EWT(1)/EWP(q)/EWT(m)/EWP(B)/BDS AFPTC/ASD/IJP(C)/SSD JD  
 ACCESSION NR: AT3002237 S/29/1/63/001/000/0290/0299

AUTHORS: Levshin, V. L.; Reshetina, T. S.; Tunitskaya, V. F.; Vasil'yeva, Ye. G. B

TITLE: Stimulating action of infrared radiation on zinc sulfide phosphors

SOURCE: Optika i spektroskopiya; sbornik statey. v. 1: Lyuminesentsiya. Moscow, Izd-vo AN SSSR, 1963, 290-299

TOPIC TAGS: electron, tr.p., energy level, infrared, absorption, flashing, phosphorescence

ABSTRACT: An investigation was made of the flashing process in ZnS with electrons trapped (or localized) in shallow levels under infrared excitation of wavelength  $1\mu$  to  $3.5\mu$ . The infrared response of these phosphors was studied at -77, -196 and -259C. Flash-emitting energy levels were established after obtaining the thermoluminescence curves of several zinc sulfide phosphors. The effect of infrared radiation of various wave lengths on one specimen, under varying conditions of excitation, was studied in great detail. It is shown that quenching, maximum absorption in radiation spectra, and the flash magnitude under stimulation of infrared radiation at the excitation level of 365 millimicron is 1.5 to 2.0 times lower than the excitation at  $\lambda = 312$  millimicron. This is attributed to action of p-type levels

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ACCESSION NR: AT3002237

(differences in trapped electron absorptions). A study was also made of the growth and decay of flashing and the phosphorescence damping at various temperatures. Orig. art. has: 7 figures and 4 tables.

ASSOCIATION: none

SUBMITTED: 29Jun62

DATE ACQ: 19May63

ENCL: 00

SUB CODE: PH

NO REF SOV: 012

OTHER: 006

Card 2/2

TUNITSKAYA, V.F.

Thermoluminescence of ZnS-phosphors in the temperature range  $14^{\circ}$  to  $77^{\circ}\text{K}$ .  
Opt. i spektr. 14 no.3:445-446 Mr '63. (MIRA 16:4)  
(Phosphors)

S/051/63/014/003/019/019  
E039/E120

**AUTHOR:** Tunitskaya, V.F.

**TITLE:** Thermal luminescence of ZnS - phosphors in the temperature range 14 - 77 °K

**PERIODICAL:** Optika i spektroskopiya, v.14, no.3, 1963, 445-446

**TEXT:** A special helium cryostat of the type described by E.N. Lotkova and A.B. Fradkov (PTE no.1, 1961, 188) is used in these experiments. The ZnS samples are in the form of a 4  $\mu$  thick layer on duralumin (1.5 mg/cm) mounted on a resistance thermometer which is tightly screwed to a cold conductor. A heating coil is situated between the sample and the helium bath, and increases the sample temperature at a rate of 9°/min. Luminescence is detected on a photomultiplier and the 365 m $\mu$  line is used for excitation. Excitation time at 14 °K is 10 minutes; the sample is then heated after a period of one minute from the time of completion of excitation. ZnS phosphors of various types are investigated. It is shown that the shape of the thermal luminescence curve is independent of the presence of copper in the range 14 - 80 °K.

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Thermal luminescence of ZnS - ...

S/051/63/014/003/019/019  
E039/E120

and shows a weak peak at about 70 °K. Similarly the curve for ZnS - Cu heated in molten  $\text{MgCl}_2$ ,  $\text{BaCl}_2$  and  $\text{ZnCl}_2$  shows the same typical shape. ZnS - Ag heated in an atmosphere of HCl produced a broader peak with a maximum at about 60 °K. These results show that the ZnS phosphors have a small trapping level. Samples with a hexagonal modification have a background extending down to 14 °K. This effect will be the subject of a further paper. At the present time it is assumed that the small levels are not connected with any activator, but depend only on the structure of the basic material. There is 1 figure.

SUBMITTED: October 20, 1962

Card 2/2

LEVSHIN, V.L.; TUNITSKAYA, V.F.

Luminescence of ZnS-Mn phosphors during excitation and its kinetics.  
Opt. 1 spektr. 9 no.2:223-232 Ag '60. (MIRA 13:8)  
(Luminescence) (Phosphors)

LEVSHIN, V.L.; TUNITSKAYA, V.F.

Thermoluminescence and localization levels of ZnS-Mn phosphors.  
Opt. i spektr. 8 no.5:663-671 My '60. (MIRA 13:9)  
(Zinc sulfide) (Luminescence)

82949

S/051/60/008/005/011/027  
E201/E491

24.3500  
AUTHORS:

Levshin, V.L. and Tunitskaya, V.F.

TITLE:

Thermal De-excitation and Localization Levels of  
ZnS-Mn Phosphors

PERIODICAL: Optika i spektroskopiya, 1960, Vol.8, No.5, pp.663-671

TEXT: A series of ZnS-Mn phosphors was studied in a wide range of temperatures using exciting light of various wavelengths and intensities. The first part of this investigation is reported in the present paper which deals with the structure, population and thermal de-excitation (emptying) of capture levels in ZnS-Mn crystals of cubic modification. The effect of increasing manganese concentration on the spectrum of the levels was studied, as well as the effect of the excitation wavelengths on the population of the levels. The phosphors were prepared by heating a charge in a stream of nitrogen at 850°C for 30 min. Manganese was introduced in the form of MnCl and its initial concentration in the charge was from  $10^{-6}$  to  $3 \times 10^{-2}$  g/g. About 40% of Mn was lost during firing. X-ray diffraction studies showed that all samples had sphalerite structure. Some of the results are presented in

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S/051/60/008/005/011/027  
E201/E491

# Thermal De-excitation and Localization Levels of ZnS-Mn Phosphors

Fig.1 to 7 and Tables 1 to 3. Fig.1 and 2 show the thermoluminescence (thermal de-excitation) curves for phosphors prepared at various temperatures (Fig.1) and with various amounts of manganese (Fig.2). Fig.3 and 4 and Tables 1 and 2 give the analysis of peaks I, II and III in thermoluminescence curves. Fig.5 and 6 show the light sums of blue, manganese and total luminescence as a function of the concentration of manganese. Fig.7 shows the thermoluminescence curves of ZnS (Fig.7a) and of ZnS with  $10^{-3}$  g/g Mn (Fig.7b) excited with light of 312 (curves 1), 366 (curves 2) and 436 mμ (curves 3). The reduction of thermoluminescence light sums when the phosphors were excited with light of 436 mμ wavelength (compared with excitation using 312 and 366 mμ light) is illustrated in Table 3. The following conclusions were drawn from the results: (1) Manganese itself does not produce new localization levels which would appear in thermoluminescence peaks above 80°K. Only a small number of Mn ions remains near places where electrons are localized by oxygen and these Mn ions deepen the oxygen localization levels producing a new peak at -30°C.

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# Thermal De-excitation and Localization Levels of ZnS-Mn Phosphors

Formation of this new peak is accompanied by a reduction of the oxygen peak at  $-90^{\circ}\text{C}$ . (2) The quantum light sums of ZnS-Mn with up to  $6 \times 10^{-3}$  g/g Mn are practically independent of the concentration of Mn. In such phosphors radiationless transitions of electrons from localization levels do not occur on thermal de-excitation. (3) Light of 312 and 366 mμ wavelengths does not affect localized electrons and produces similar light sums on thermal de-excitation. If light of  $\lambda = 436$  mμ is used, the light sums are reduced by factor of 3 and this reduction is stronger in the case of the blue luminescence compared than for orange luminescence; luminescence due to shallow levels is weakened to a greater extent than luminescence due to deep levels. (4) Increase of the amount of Mn reduces the depth of electron localization levels. (5) The profile of the high-temperature thermoluminescence peak suggests that only infrequently electrons are recaptured by localization levels during the process of thermoluminescence. Acknowledgments are made to A.V.Lavrov for preparation and analysis of the phosphors and to O.M.Agafonova for

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S/051/60/008/005/011/027  
E201/E491

Thermal De-excitation and Localization Levels of ZnS-Mn Phosphors  
her help in experiments. There are 7 figures, 3 tables and  
13 references: 11 Soviet, 1 German and 1 Dutch.

SUBMITTED: July 25, 1959

✓

Card 4/4

L 34871-65 EWT(1)/EWT(n)/EW(t)/EWP(b) IJP(c) JD

ACCESSION NR: AP5005051

8/0051/65/018/002/0328/0730

AUTHOR: Levshin, V. L.; Tunitskaya, V. F.

TITLE: On some luminescence characteristics of ZnS-Mn phosphors in the 15-77K region

TOPIC TAGS: luminescence, thermoluminescence, zinc sulfide optic material, low temperature research

ABSTRACT: The authors have extended their earlier researches on the luminescent

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S/051/60/009/002/002/008  
E201/E691

AUTHORS: Levshin, V.L. and Tunitskaya, V.F.

TITLE: The Luminescence Processes and their Kinetics in ZnS-Mn Phosphors  
During Excitation

PERIODICAL: Optika i spektroskopiya, 1960, Vol. 9, No. 2, pp. 223-232

TEXT: In an earlier paper (Ref 1) the authors dealt with thermal deexcitation and the origin of localization levels in ZnS-Mn phosphors. The present paper extends this work to luminescence of the same phosphors under excitation. The phosphors were in the form of  $\sim 3.7 \mu$  thick layers (with grain diameter of  $\sim 1 \mu$ ) and were excited with 365 and 312 m $\mu$  lines from a mercury lamp; the quantum intensities of the lines were equalized with filters. The intensity of luminescence was measured with a photomultiplier FEU-32. The author studied the effect of temperature on the absorption spectra (Figs. 1, 2) and on the blue and orange luminescence of the phosphors (Figs. 3-6). The information obtained is used to discuss the processes of transformation of the excitation energy into blue luminescence, manganese luminescence and heat, as a function of the ambient temperature and the concentration of manganese.

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S/051/60/009/002/002/003

E201/E691

The Luminescence Processes and their Kinetics in ZnS-Mn Phosphors During  
Excitation

The mechanism of luminescence and quenching processes are dealt with and an energy diagram is proposed (Fig. 7) for the localization levels of electrons and holes. Acknowledgments are made to O.M. Agafonova for her help in measurements and to A.V. Lavrov for preparation of the phosphors. There are 7 figures, 1 table and 11 Soviet references. ✓

SUBMITTED: December 17, 1959

Card 2/2

TUMITSKAYA, V.F.

USSR / Optics

K

Abs Jour: Referat Zhur-Fizika, 1957, No 4, 10379

Author : Levshin, V.L., Tumitskaya, V.F., Cherepneva, A .A.  
Inst : Physics Institute, Academy of Sciences, USSR  
Title : Origin of Localization Levels in ZnS-Cu and Co Phosphors.

Orig Pub: Optika i spektrokopiya, 1956, 1, No 2, 255-263

Abstract: An investigation was made of the thermal glow (TG) of the phosphors ZnS, ZnS-Cu, ZnS-Co and ZnS-(Cu, Co) (annealing in  $H_2S$  and air for 30 minutes). In ZnS, the azure glow (bands at approximately 460 millimicrons) occurs only in the presence of a flux ( $CaCl_2$ ). The peak of TG at  $-130^\circ$  is due to the superstoichiometric zinc (for which favorable circumstances are produced by the chlorine), and the peak at  $-60^\circ$  is ascribed to the oxygen. The green glow is ascribed to traces of copper. In ZnS-Cu, in addition to the zinc and oxygen peaks, there appear three new peaks, barely noticeable at  $-5$  and  $0^\circ$  and a considerable one at  $+20^\circ$ . These are ascribed to copper and appear to be the cause of the longer

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USSR / Optics

K

Abs Jour: Referat Zhur-Fizika, 1957, No 4, 10379

afterglow of ZnS-Cu at room temperature. ZnS-Co during the instant of excitation has an azure glow, weak at room temperature, intense at  $-186^{\circ}$ . Introducing Co decreases sharply the light sums from the small local levels. Simultaneously there appear deeper levels in the region of  $+50^{\circ}$ . Their structure is not clear because of the small light sums. In ZnS-(Cu, Co), the peaks at  $-130$  and  $-60^{\circ}$  are suppressed rapidly with increasing Co, and the copper peaks ( $-5, 0$ , and  $+20^{\circ}$ ) are suppressed slowly, while new peaks appear at  $50$  and  $80^{\circ}$ .

Card : 2/2

GUDKOV, A.; BUSURIN, Ya.; IOFE, N.; PALKIN, G., kand. sel'khoz. nauk;  
TUNITSKIY, A., red.; KOROTAYEVA, D., tekhn. red.

[Manual on private livestock and poultry raising] Spravochnik  
po individual'nomu zhiivotnovodstvu i ptitsevodstvu. Moskva,  
Izd-vo VTsSPS Profizdat, 1946. 182 p. (MIRA 14:8)  
(Stock and stockbreeding) (Poultry)

TUNITSKIY, A.

Bryl', Yanka

"Dawn breaks in Zabolot'ye." Yanka Bryl', Reviewed by A. Tunitskiy, Mol. kolkh. 19,  
No. 9, 1952.

Monthly List of Russian Accessions, Library of Congress, December 1952. Unclassified.

TUNITSKIY, A.

White Russian Literature - History and Criticism

"Dawn breaks in Zabolot'ye." Ianka Bryl'. Reviewed by A. Tunitskiy. Mol.kolkh.  
19 No. 9, 1952

Monthly List of Russian Accessions, Library of Congress, December 1952. Unclassified.

TUNITSKIY, L.N., kandidat fiziko-matematicheskikh nauk; IGHASHKOV, A.I.,  
~~kandidat fiziko-matematicheskikh nauk~~

Electric discharge lighting in an elongated tube. Svetotekhnika  
1 no.2:23-26 Ap '55. (MLRA 8:9)

1. Moskovskiy elektrolampovyy zavod.  
(Fluorescent lighting)

SKOBELEV, V.M., inzhener; TUNITSKIY, L.N., kandidat fiziko-  
matematicheskikh nauk

Investigation of factors determining the start in fluorescent  
lamps. Svetotekhnika 1 no.4:14-17 Ag '55. (MLRA 8:9)

1. Moskovskiy elektrolampovyy zavod.  
(Fluorescent lamps)

PRIKHOT'KO, A.F.

24(7)

3

PHASE I BOOK EXPLOITATION SOV/1365

L'vov. Universitet

Materialy I Vsesoyuznogo soveshchaniya po spektroskopii. t. 1: Molekulyarnaya spektroskopiya (Papers of the 10th All-Union Conference on Spectroscopy. Vol. 1: Molecular Spectroscopy) [L'vov] Izd-vo L'vovskogo univ-ta, 1957. 499 p. 4,000 copies printed. (Series: Its: Fizichnyy sbirnyk, vyp. 3/8/)

Additional Sponsoring Agency: Akademiya nauk SSSR. Komissiya po spektroskopii. Ed.: Jazer, S.L.; Tech. Ed.: Saranyuk, T.V.; Editorial Board: Landsterg, G.S., Academician (Resp. Ed., Deceased), Neporent, B.S., Doctor of Physical and Mathematical Sciences, Pabelinskiy, I.L., Doctor of Physical and Mathematical Sciences, Fabrikant, V.A., Doctor of Physical and Mathematical Sciences, Kornitskiy, V.G., Candidate of Technical Sciences, Rayskiy, S.M., Candidate of Physical and Mathematical Sciences, Klimovskiy, L.K., Candidate of Physical and Mathematical Sciences, Milyanchuk, V.S., Candidate of Physical and Mathematical Sciences, and Glauberman, A. Ye., Candidate of Physical and Mathematical Sciences.

Card 1/30

Rutyrkin, V.N., Sh. I. Paygulayev, and L.N. Tunitskiy. Study of Spectrum of B<sub>2</sub>

486

Zelenskaya, L.G., and L. N. Tunitskiy. Study of the Spectrum of a BF<sub>3</sub> VACUUM DISCHARGE in the Ultra-violet

489

Veselago, V.G., and A.M. Prokhorov. Micro-wave Spectrum of a HDS<sub>2</sub> Molecule

493

AVAILABLE: Library of Congress

TN/rj  
4-22-59

Card 30/30

PHASE I BOOK EXPLOITATION SOV/1297

Vsesoyuznaya nauchno-tekhnicheskaya konferentsiya po primeneniyu radioaktivnykh i stabil'nykh izotopov i izucheniya v narodnom khozyaystve i nauke, Moscow, 1957

Polucheniye izotopov. Mozhnyye gamma-izotopi. Radiometriya i dosimetriya. Trudy konferentsii... (Isotope Production. Radiometry. Transacti... of the All-Union Conference on the Use of Radioactive and Stable Isotopes and Radiation in the National Economy and Science, Moscow, Izd-vo AN SSSR, 1958. 293 p.

Sponsoring Agency: Akademiya nauk SSSR; Glavnoye upravleniye po ispol'zovaniyu atomnoy energii SSSR.

Editorial Board: Prolov, Yu.S. (Resp. Ed.), Zhavoronkov, M.M. (Deputy Resp. Ed.), Agintsev, K.K., Aleksyev, B.A., Bochkarev, V.V., Lushchinsky, M.I., Malkov, T.P., Sintsev, V.I., and Popov, G.L. (Secretary); Tech. Ed.: Novikov, M.D.

FOURTH: This collection is published for scientists, technologists, persons engaged in medicine or medical research, and others concerned with the production and/or use of radioactive and stable isotopes and radiation.

CONTENTS: Thirty-eight reports are included in this collection under three main subject divisions: 1) production of isotopes; 2) high-energy gamma-radiation facilities, and 3) radiometry and dosimetry.

TABLE OF CONTENTS:

PART I. PRODUCTION OF ISOTOPES

Prolov, Yu.S., V.V. Bochkarev, and Ye.Ye. Kulish. Development of Isotope Production in the Soviet Union. This report is a general survey of production methods, apparatus, raw materials, applications, investigations and future prospects for radio isotopes in the Soviet Union. Card 2/12

Kyulankordt, Yu.K., G.D. Zivert, and T.A. Gagua. A Rectification Column for Obtaining HF<sub>3</sub> Enriched With Isotope P<sub>32</sub>. A method is described for enriching natural mixtures containing ~18.6 percent B<sub>10</sub> concentration to ~80 percent B<sub>10</sub> concentration by low temperature (~ -100 degrees, scale not stated) and azeotropic rectification. Separation capability was B<sub>10</sub> of 95-96 percent purity after 80 hours processing; but, as the desired concentration was ~80 percent, separation yield was 4 liters per 24 hours. Block diagrams of installations are given. 127

Zhavoronkov, M.M., O.V. Uvarov, and S.I. Babkov. Research on the Separation of Stable Isotopes of Light Elements. Dmititskiy, M.M., G.D. Derzhatyn, M.V. Tikhomirov, and A.D. Zorin, and M.L. Nikolayev. Separation of Carbon Isotopes. 134

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143



SOV/51-5-5-5/23

AUTHORS: Tatevskiy, V.M., Tunitskiy, L.M., and Novikov, M.M.

TITLE: Vibrational Constants and Dissociation Energy of the BeF Molecule.  
(Kolebatel'nyye postoyannyye i energiya dissotsiatsii molekuly BeF)

PERIODICAL: Optika i Spektroskopiya, 1958, Vol 5, Nr 5, pp 520-529 (USSR)

ABSTRACT: The paper reports a new investigation of the vibrational structure of the  $A^2\Pi-X^2\Sigma$  bands of BeF. The BeF spectrum was excited using gas-discharge tubes. Since the gas temperature is comparatively low in such tubes the rotational structure of the bands was weaker than in an arc and this reduced overlapping of bands and made it easier to observe new band edges. Two types of tubes were used in this work: one working under steady-state conditions (Fig 1) and the other for pulse excitation (Fig 2). Both tubes were filled with helium under 2-3 mm Hg pressure. BeF molecules were introduced into the discharge tubes by placing some BeF<sub>2</sub> in nickel boats inside the tubes. The tube used for steady-state discharges was U-shaped (1, in Fig 1) and had a quartz window (7, in Fig 1). The space around the electrodes (3, in Fig 1) was joined to the tube proper via liquid-hydrogen traps (4 and 5 in Fig 1). The tube was supplied from a 630 W transformer at 13 kV. The pulse-discharge source

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SOV/51-5-5-5/23

Vibrational Constants and Dissociation Energy of the BeF Molecule

consisted of a straight tube (1, in Fig 2) with a fluorite window (2, in Fig 2). Its electrodes (4, in Fig 2) were supplied every second (for 0.02 sec) with a 600-700 V, 9 A pulse. The spectra were recorded using a DFS-3 spectrograph of 2 Å/mm dispersion and 144000 resolving power. An iron spectrum was used as a wavelength standard. Measurements were made with 0.5 cm<sup>-1</sup> precision. The spectrum is shown in Fig 3. The measured band edges are given in Table 1. 48 new edges of Q<sub>1</sub>-branches and 54 new edges of R<sub>2</sub> and R<sub>1</sub> branches were recorded. Table 2 gives the rotational constants  $\omega_0$ ,  $\omega_0 x_0$  and  $\omega_0 y_0$  of BeF taken from Refs 1, 3, 5, 6 and from the results reported in the present paper. Table 3 gives the values of  $v'_{\max}$ ,  $v_{\max}$ ,  $G'_0(v'_{\max})$  and  $G''_0(v_{\max})$  allowing for (columns I) and neglecting (columns II) the second coefficient of anharmonicity  $\omega_0 y_0$  of the BeF molecule. Table 4 gives the recommended, most reliable values of the spectroscopic constants of the BeF molecule

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SOV/51-5-5-5/23

Vibrational Constants and Dissociation Energy of the BeF Molecule

This table gives the dissociation energies of the  $X^2\Sigma$  and  $A^2\Pi$  states as  $8 \pm 0.5$  and  $3.9 \pm 0.5$  eV respectively. The author thanks L.V. Gurvich for his advice. There are 4 figures, 4 tables and 7 references.

SUBMITTED: December 6, 1957

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1. Beryllium fluoride--Spectrographic analysis 2. Molecules--Spectra  
3. Molecules--Energy 4. Gas discharges--Applications

30541  
S/051/60/008/06/002/024  
E201/E691

5.4/30

AUTHORS: Ovcharenko, I.Ye., Tunitskiy, L.N. and Yakutin, V.I.

TITLE: Analysis of the Fine Structure of the SiCl Molecular Bands

PERIODICAL: Optika i spektroskopiya, 1960, Vol. 8, Nr 6, pp 746-751 (USSR)

ABSTRACT: Four electron states ( $X^2\Pi$ ,  $B^2\Sigma$ ,  $C^2\Delta$  and  $D^2\Sigma$ ) of the SiCl molecule are known (Refs 1-4). The vibrational constants of these four states were reported by Jevons (Ref 4) and are listed in Table 1; the constants with question marks were considered unreliable by Jevons. The present paper reports new values of the rotational constants of the  $B^2\Sigma$ ,  $X^2\Pi_{3/2}$ ,  $X^2\Pi_{1/2}$  states, derived from the rotational analysis of the (1, 0), (0, 0) and (0, 1) bands of the  $B^2\Sigma \rightarrow X^2\Pi$  transition. The spectra of SiCl were excited in a quartz pulse-discharge tube, similar to one used earlier (Ref 5) and shown schematically in a figure on p 746. The tube was filled with a mixture of silicon tetrachloride and helium. The spectra were photographed with a DFS-3 spectrograph in the third order (dispersion of 0.57 Å/mm, resolving power of 432 000), and measured with a IZA-2 comparator (an iron arc spectrum was used as the wavelength standard, cf. Table 2).

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S/051/60/008/06/002/024  
E201/E691

Analysis of the Fine Structure of the SiCl Molecular Bands

For the  $X^2\Pi_{1/2}$  state the following rotational constants were obtained:  $B_1 = 0.2550$ ,  $\alpha = 0.0016$ ,  $D_e = 2.341 \times 10^{-7}$ ,  $a = 0.004 \pm 0.001 \text{ cm}^{-1}$ ; for  $X^2\Pi_{3/2}$ :  $B_2 = 0.2556$ ,  $\alpha = 0.0016$ ,  $D_e = 2.355 \times 10^{-7} \text{ cm}^{-1}$ ; for  $B^2\Sigma$ :  $B = 0.2782$ ,  $\alpha = 0.0015$ ,  $D_e = 1.752 \times 10^{-7} \text{ cm}^{-1}$  (Table 3). The wavenumbers of the (0.1), (0.0), (1.0) lines of the  $2\Sigma \rightarrow 2\Pi_{1/2}$  transition were respectively 33662.0, 34193.6, 34892.2  $\text{cm}^{-1}$ ; for the  $2\Sigma \rightarrow 2\Pi_{3/2}$  they were 33455.7, 33987.1, 34685.8  $\text{cm}^{-1}$ , respectively. There are 1 figure, 3 tables and 6 references, of which 2 are Soviet, 3 English and 1 German.

SUBMITTED: July 20, 1959

Card 2/2

5.4/30  
S/051/60/008/06/003/014  
E201/E691

AUTHORS: Novikov, M.M. and Tunitskiy, L.N.

TITLE: The Vibrational Constants and the Dissociation Energy of the BeCl Molecule

PERIODICAL: Optika i spektroskopiya, 1960, Vol 8, Nr 6, pp 752-760 (USSR)

ABSTRACT: Two electron states ( $X^2\Sigma$  and  $A^2\Pi$ ) of the BeCl molecule are known. Fredrickson and Hogan (Ref 1) determined the vibrational constants,  $\omega_e$  and  $\omega_e x_e$ , and estimated the rotational constants  $B'_0$  and  $B''_0$ . Fredrickson and Hogan (Ref 1) used linear approximation to obtain the dissociation energies of the  $X^2\Sigma$  (4.33 eV) and  $A^2\Pi$  (3.47 eV) states. Gaydon (Ref 3) corrected the former value to  $3 \pm 0.5$  eV. The present paper reports a new investigation of the vibrational structure of the BeCl molecular band. A quartz pulse-discharge tube (Ref 4) was employed. The spectra (cf. a figure on p 753) were photographed with a DFS-3 spectrograph (dispersion of 2 Å/mm, resolving power of 144 000). The spectra were measured with a comparator IZA-2 (an iron arc spectrum was used as the wavelength standard). Thirty new  $Q_1$  band edges and 43 new  $R_1$  and  $R_2$  band edges were recorded. More precise values (Table 6) of the

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S/051/60/008/06/003/024

E201/E691

## The Vibrational Constants and the Dissociation Energy of the BeCl Molecule

vibrational constants  $\omega_0$  and  $\omega_0 x_0$  of the  $X^2\Sigma$  (841.3 and 5.11  $\text{cm}^{-1}$  respectively) and the  $A^2\Pi$  (816.0 and 5.06  $\text{cm}^{-1}$  respectively) states were obtained. The second coefficients of anharmonicity  $\omega_0 y_0$  were calculated for both states (0.0205 for  $X^2\Sigma$  and -0.0368  $\text{cm}^{-1}$  for  $A^2\Pi$ ). The third coefficient of anharmonicity of the  $X^2\Sigma$  state was found to be -0.0000582  $\text{cm}^{-1}$ . Non-linear extrapolation yielded a new and more reliable value of the dissociation energy of BeCl;  $5.9 \pm 0.5$  eV (Table 6). In the 2610-2620 Å region new bands were discovered (Table 7) which are due to a transition from a hitherto unknown electron state. There are 1 figure, 6 tables and 6 references, of which 2 are Soviet, 2 English, 1 French and 1 translation from English into Russian.

SUBMITTED: September 21, 1959

Card 2/2.

KUZYAKOV, Yu.Ya.; TATEVSKIY, V.M.; TUNITSKIY, L.N.

Rotational analysis of boron monoxide bands located in the vacuum  
ultraviolet region. Opt. i spektr. 9 no.2:156-161 Ag '60.  
(MIRA 13:8)

(Boron oxide--Spectra)



TUNITSKIY, L. N.

95  
S/089/62/013/006/019/027  
B102/B186

AUTHORS: G. T. and M. R.

TITLE: Nauchnaya konferentsiya Moskovskogo inzhenerno-fizicheskogo instituta (Scientific Conference of the Moscow Engineering Physics Institute) 1962

PERIODICAL: Atomnaya energiya, v. 13, no. 6, 1962, 603 - 606

TEXT: The annual conference took place in May 1962 with more than 400 delegates participating. A review is given of these lectures that are assumed to be of interest for the readers of Atomnaya energiya. They are following: A. I. Leypunskiy, future of fast reactors; A. A. Vasil'yev, design of accelerators for superhigh energies; I. Ya. Pomeranchuk, analyticity, unitarity, and asymptotic behavior of strong interactions at high energies; A. B. Migdal, phenomenological theory for the many-body problem; Yu. D. Fivyskiy, deceleration of medium-energy antiprotons in matter; Yu. M. Kogan, Ya. A. Iosilevskiy, theory of the Mössbauer effect; M. I. Ryazanov, theory of ionisation losses in nonhomogeneous medium; Yu. B. Ivanov, A. A. Rukhadse, h-f conductivity of subcritical plasma;

Card 1/4

36

S/089/62/013/006/019/027  
B102/B186

**Nauchnaya konferentsiya...**

Ye. Ye. Lovetskiy, A. A. Rukhadze, electromagnetic waves in nonhomogeneous plasma; Yu. D. Kotov, I. L. Rozental', the origin of fast cosmic muons; Yu. M. Ivanov, muon depolarization in solids; V. G. Varlamov, Yu. M. Grashin, B. A. Dolgoshein, V. G. Kirillov-Ugryumov, V. S. Roganov, A. V. Samoylov,  $\mu^-$  capture by various nuclei; V. S. Demidov, V. G. Kirillov-Ugryumov, A. K. Ponomarev, V. P. Protasov, F. M. Sergeyev, scattering of  $\pi^-$  mesons at 5 - 15 Mev in a propane bubble chamber; S. Ya. Nikitin, M. S. Aynutdinov, Ya. M. Selektor, S. M. Zombkovskiy, A. P. Grashin, muon production in  $\pi^+p$  interactions; B. A. Dolgoshein, spark chambers; N. G. Volkov, V. K. Lyapidevskiy, I. M. Obodovskiy, study of operation of a convection chamber; K. G. Finogenov, production of square voltage pulses of high amplitudes; G. N. Alekseev, problems of color vision; V. K. Lyapidevskiy, relation between number of receivers and number of independent colors; Ye. M. Kudryavtsev, N. N. Sobolev, N. I. Tizengauzen, L. N. Tunitskiy, F. S. Fayzulin, determination of the moment of electron transition of oscillator forces and the widths of the Schumann-Runge bands of molecular oxygen; B. Ye. Gavrilov, A. V. Zharikov, V. I. Rayko, decomposition of the volume charge of intense ion beams; Ye. A. Kramer-Ageyev, V. S. Troshin, measurement of neutron spectra; G. G. Doroshenko, new methods of fast-neutron recording; V. I. Ivanov, dosimetry terminology; R. M. Voronkov, Card 2/4.

KUDRYAVTSEV, Ye. M.; SOBOLEV, N.N.; TUNITSKIY, L.N.; FAYZULLOV, F.S.

Pyrometric study of the state of a gas behind a reflected  
shock wave. Trudy Fiz.inst. 18:159-200 '62. (MIRA 15:12)  
(Pyrometry) (Shock waves) (Gas dynamics)

S/051/63/014/003/005/019  
E039/E120

AUTHORS: Krindach, N.I., Kudryavtsev, Ye.M., Sobolev, N.N.,  
Tunitskiy, L.N., and Fayzulloev, F.S.

TITLE: Determination of the electronic transition moments of  
the Schumann-Runge band system in oxygen. I.

PERIODICAL: Optika i spektroskopiya, v.14, no.3, 1963, 351-361

TEXT: A method is developed for determining the squares of  
matrix elements for electron transitions of molecules based on a  
measurement of the absorption in gases behind shock waves. The  
Schumann-Runge absorption bands for oxygen are obtained on a shock  
tube using helium as a working gas and equipped with apparatus for  
measuring temperature by the usual method of rotation of spectral  
lines; in this case by observing the resonance line of  
Ba II ( $\lambda = 4554 \text{ \AA}$ ). An ionization probe is used for measuring the  
velocity of the front of the incident shock waves. The pulsed  
light source and the synchronizing circuit are described in  
detail. Spectra are recorded on ДФС-13 (DFS-13) and KC-55  
(KS-55) spectrographs with glass and with quartz optics. By  
plotting the absorption index against wave number  $\sim$  for the  
Card 1/2

Determination of the electronic ... S/051/63/014/003/005/019  
E039/E120

groups  $R(K)P(K - 4)$  and measuring the area under the curve, a value for the integral of the absorption index is obtained for lines of R and P form, from which is calculated the square of the moment for electronic transitions  $|R_e^m|^2$ . The value of this integral obtained graphically agrees with the calculated value. These results and the analysis of possible errors will be examined in a later paper. There are 10 figures.

SUBMITTED: May 18, 1962

Card 2/2

KRINDACH, N.I.; SILIN-BEKCHURIN, I.A.; TUNITSKIY, L.N.; CHERKASOV, Ye.M.

High-frequency discharge in a neon-helium laser. Zhur. tekhn. fiz.  
35 no.9:1678-1684 S '65. (MIRA 18:10)

1. Fizicheskiy institut imeni P.N.Lebedeva AN SSSR, Moskva.

L 2983-66 EWA(k)/FED/EWT(1)/EWT(m)/EPF(c)/EEC(k)-2/T/EWP(t)/EWP(k)/EWP(b)/  
 ACCESSION NR: AP5024051 EWA(m)-2/EWA(h) SCTB/ UR/0057/65/035/009/1678/1684  
 WG/JD LJP(c) 537.523.7 53  
 51  
 8  
 AUTHOR: Krindach, N. I.; Silin-Bekchurin, I. A.; Tunitskiy, L. N.; Cherkasov, Ye. M.  
 TITLE: Study of a high-frequency discharge in a neon-helium laser 44  
 SOURCE: Zhurnal tekhnicheskoy fiziki, v. 35, no. 9, 1965, 1678-1684 27  
 TOPIC TAGS: gas laser, neon helium laser, hf discharge, plasma discharge 25, 44  
 ABSTRACT: A new method is proposed for determining the current and voltage distribution along a high-frequency discharge and a study is made of the effect on laser operation of inhomogeneities along such a discharge. The method is based on the assumption that the voltage and current at any cross section of a discharge tube can be determined by the distance of that cross section from the end of the glowing portion of discharge. This assumption holds for any stationary discharge at any cross section of which electron rise due to ionization is a unity. The experiments were carried out by means of a gas laser ( $\lambda = 6328 \text{ \AA}$ ) (see Fig. 1 of the Enclosure) which incorporated a fused-quartz discharge tube 1.7 cm long and 8 mm in diameter (internal) filled with a neon — helium mixture at a 10:1 ratio at a pressure of 0.8 mm Hg. Two plane-parallel quartz plates  $O_1$  and  $O_2$  were  
 Cont 1/4

L 2983-66

ACCESSION NR: AP5024051

2

placed at the tube ends at Brewster's angle. The equivalent circuit of the discharge tube is shown in Fig. 2. The tube was placed between the two confocal dielectric mirrors M with a 2-m radius of curvature and an ~ 1% transmission around 6328 Å. The mirrors were adjusted by means of an AKT-400 collimator. The discharge tube was fed by a 30-Mc frequency from an h-f oscillator, whose voltage was supplied to 8-cm electrodes  $E_1$  and  $E_2$ , while electrodes  $E_3$ ,  $E_4$ , and  $E_5$  (2.5 cm each) were grounded. The oscillator was L-coupled to the discharge tube and the currents  $I_1$  and  $I_2$  and voltages  $V_1$  and  $V_2$  were measured by T-22 hot-wire ammeters and S-95 electrostatic voltmeters (4-pf input capacitance) respectively. The output energy was measured by means of a calibrated thermopile. The capacity of the discharge tube, varied by a movable ground rod R placed above the tube, was determined by its distance from R. In the experiments a discharge with a maximum length of 35 cm was studied. The experimental method and results are discussed in detail and indicate good agreement with computed data. Orig. art. has: 1 table and 7 figures.

[YK]

ASSOCIATION: Fizicheskii institut imeni P. N. Lebedeva AN SSSR, Moscow (Physics Institute, AN SSSR)

44

SUBMITTED: 18Jan65

ENCL: 02

SUB CODE: EC

NO REF SOV: 004

OTHER: 006

ATD PRESS 4/1/66



L 2983-66

ACCESSION NR: AP5024051

ENCLOSURE: 01

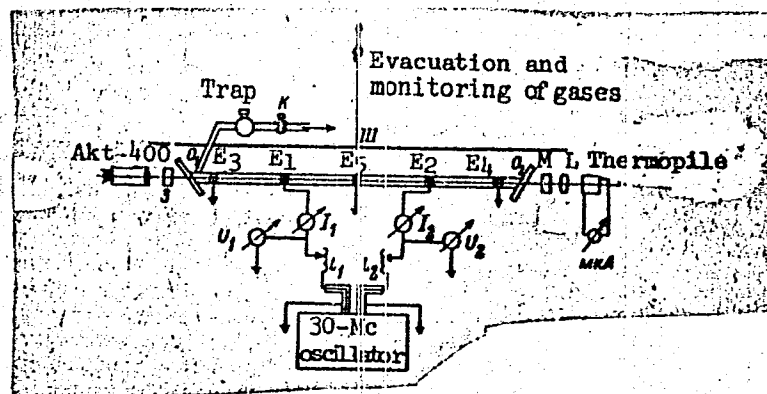


Fig. 1. Schematic of the laser

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L 2983-66

ACCESSION NR: AP5024051

ENCLOSURE: 02

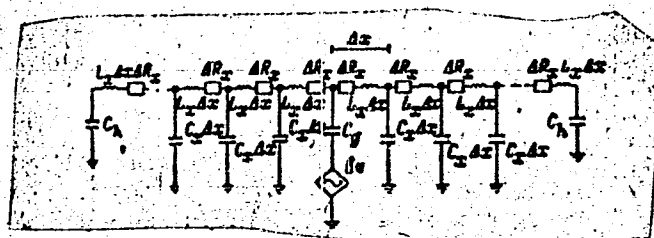


Fig. 2. Equivalent circuit of the discharge tube

BVK  
Card 4/4

*TUNITSKIY, L.M.*  
L 46301-65 EWT(1) IJP(c)

UR/0058/65/000/003/D013/D013

ACCESSION NR: AR5012225

SOURCE: Ref. zh. Fizika, Abs. 3D81

AUTHOR: Sobolev, N.N.; Antropov, Ye.T.; Gippius, Ye.F.; Dronov, A. P.; Krindach, M.L.; Kudryavtsev, Ye.M.;  
Pechenov, A.N.; Sviridov, A.G.; Tunitkiy, L.M.; Fayzullov, F.S.; Cherepishnikov, V. P.

TITLE: Experimental determination of electronic oscillator strengths of diatomic molecules

CITED SOURCE: Tr. Komis. po spektroskopii. AN SSSR, vyp. 1, 1964, 64-81

TOPIC TAGS: oscillator strength, electron oscillator, diatomic molecule, shock wave, oxygen, nitric oxide, cyan, electronic spectrum

TRANSLATION: To determine the oscillator strengths of electronic transitions of diatomic molecules, an experimental method was developed, based on the measurement of the absorption of gas behind a shock wave reflected from the end of a shock tube. By varying the velocity of the incident shock wave and by calculating the state of the gas behind the shock wave, it is possible to determine the temperature and the concentration that the molecules behind the reflected

Card 1/2

L 46301-65

ACCESSION NR: AR5012225

Shock wave must have to permit determination of the oscillator strengths from the measured absorption. The theoretically obtained temperature was monitored by two experimental methods. A method was also developed for determining the oscillator strengths from the study of the gas behind the shock wave; these strengths were determined for the Schumann-Runge bands of oxygen, the beta and gamma systems of nitric oxide, the violet band system of OH, and the C<sub>2</sub> Swan bands.

SUB CODE: NF, ME

ENCL: 00

Card 2/2

L 64720-65

ACCESSION NR: AR5012274

UR/0058/65/000/003/D046/D046

SOURCE: Ref. zh. Fizika, Abs. 3D348

AUTHOR: <sup>44</sup> ~~Prigodan, N. I.; Sobolev, M. M.; Guritskiy, L. N.~~

TITLE: <sup>44</sup> ~~Method for determining the width of rotational lines and integral indices~~

CITED SOURCE: In. Komis. po spektroskopii. AN SSSR, vyp. 1, 1964, 704-712

TOPIC TERM: line spectrum, absorption spectrum, molecular spectrum, spectrographic analysis, absorption coefficient

TRANSLATION: A method is developed for determining the true values of the half-width of a rotational line and the integral indices. This method is applicable to the spectral instrument. The method is applicable to the case where there is incomplete resolution of the rotational structure. This method is also used to estimate the error in the integral absorption index which is caused by substituting a contour with "detached" lines for the "complete" contour. The contour which is defined in an infinite interval.

SUB CODE: OF  
Card 1/1 000

ENCL. 1

KRINDACH, N.I.; SOBOLEV, N.N.; TUNITSKIY, L.N.

Determination of the electron transition moments of the Schumann-  
Runge bands of the oxygen molecule. Part 3. Opt. i spektr. 15  
no.5:601-608 N '63. (MIRA 16:12)

TUNITSKIY, L. N.

Opredeleniye momentov elektronogo perekhoda sistem polos Shumana-Runge kisloroda

report submitted for the VIIth European Congress on Molecular Spectroscopy, Budapest,  
22-27 Jul 1963.

KRINDACH, N.I.; SOBOLEV, N.N.; TUNITSKIY, L.N.

Determining the electron transition moments of the Schumann-  
Runge bands for the oxygen molecule. Part. 2. Opt. i spektr.  
15 no.3:298-305 S '63. (MIRA 16:10)



S/051/60/009/002/007/013/XX  
E201/13491

AUTHORS: Kuzyakov, Yu.Ya., Tatevskiy, V.M. and Tunitskiy, L.N.  
TITLE: A Rotational Analysis of Boron Monoxide Bands Lying in  
the Vacuum Ultraviolet Region ✓

PERIODICAL: Optika i spektroskopiya, 1960, Vol.9, No.2, pp.156-161

TEXT: Chretien (Ref.1) studied the 1300 to 2100 Å spectra of discharges in BF<sub>3</sub> with a vacuum spectrograph fitted with a grating of 1m radius and 8.3 Å/mm dispersion. Chretien found several bands which he ascribed to the BO molecule; the band edges, interpretations and relative intensities are given in Table 1. Later, Zelenskaya and Tunitskiy (Ref.2) reported a discharge spectrum of BF<sub>3</sub> recorded in the vacuum ultraviolet region with a spectrograph DFC-5 (DFS-5) with a resolving power of 120000 and a dispersion of 2.7 Å/mm. Zelenskaya and Tunitskiy wrongly ascribed the bands in the 1700 to 1900 Å region to a hydrogen compound of boron. More detailed studies showed that Chretien's interpretation was correct. In the present paper the authors analyse the fine structure of the (0, 0) and (1, 0) bands (Chretien's nomenclature). A study of the isotopic shift in the BO bands, carried out in the authors' laboratory by A.A.Mal'tsev and V.M.Tatevskiy, showed that Chretien's interpretation should be corrected so that his (0, 0) and Card 1/2 ✓

S/051/60/009/002/007/013/XX  
E201/E491

A Rotational Analysis of Boron Monoxide Bands Lying in the Vacuum  
Ultraviolet Region

(1, 0) bands become (1, 0) and (2, 0) respectively. These two bands lie between 1300 and 2100 Å and are due to  $2\Pi \rightarrow x^2\Sigma$  transitions of the BO molecule; interpretation of the (1, 0) band is given in Table 2 and of the (2, 0) band in Table 3. The rotational constants of the  $2\Pi$  state were also determined (Table 4). There are 4 tables and 7 references: 2 Soviet, 3 English, 1 Swiss and 1 translation from English into Russian.

SUBMITTED: November 23, 1959

Card 2/2

24.3400

also 2308

83636

S/081/60/000/015/002/014

A006/A001

Translation from: Referativnyy zhurnal, Khimiya, 1960, No. 15, p. 15, # 60233

AUTHORS: Kutyarkin, V.N., Peyzulayev, Sh.I., Tunitskiy, L.N.

TITLE: Investigation of the BeF Spectrum

PERIODICAL: Fiz. sb. L'vovsk. un-t, 1957, No. 3 (8), pp. 486-489

TEXT: A ДФС -3 (DFS-3) spectrograph (2A/mm dispersion) was used to investigate the BeF emission spectrum in the 2800-3400 Å range (  $\Pi$ - $\Sigma$  transition) in a discharge tube heated to 750-800°C. A reduction of the rotational structure to  $K \approx 15-18$  as compared to the arc spectrum ( $K > 55$ ), made it possible to obtain a considerably greater number of band edges than in operating with an arc. Wave numbers of 1-1 band lines are given. The presence of lines with  $K = 0$  and 1 in the branch show that the BeF molecule terms are inverted ones. See also RZhKhim, 1959, No. 8, # 26114. X

A. Mal'tsev

Translator's note: This is the full translation of the original Russian abstract.

Card 1/1

83637

5.2460A

S/081/60/000/015/003/014

A006/A001

Translation from: Referativnyy zhurnal, Khimiya, 1960, No. 15, p. 15, # 60234

AUTHORS: Zelenskaya, L.G., Tunitskiy, L.N.

TITLE: Investigation of a Discharge Spectrum in  $\text{BF}_3$  in Vacuum Ultraviolet

PERIODICAL: Fiz. sb. L'vovsk. un-t, 1957, No. 3 (8), pp. 489-493

TEXT: АИФ-5 (DFS-5) vacuum spectrograph (2.7 Å/mm dispersion) was used to investigate the discharge emission spectrum in  $\text{BF}_3$ . The rotational structure of two bands in the 1700-1900 Å range was determined. The bands must according to Cretenin's data (Cretenin, Helv. phys. acta, 1950, Vol. 23, p. 259) belong to the  $^2\Pi - ^2\Sigma$  transition of the BO molecule, appearing as a contamination. As a result of rotational analysis it is shown that it is probably wrong to relate these bands to the BO molecule, and they are related to the  $^1\Pi - ^1\Pi$  or  $^2\Sigma - ^2\Sigma$  transition of some hydrogen compound. Rotational constants  $B'$  and  $B'' \approx 8.7 \text{ cm}^{-1}\text{Å}$  are determined.

A. Mal'tsev

Translator's note: This is the full translation of the original Russian abstract.

Card 1/1

TUNITSKIY, N. (Odessa)

First Soviet jet fighter. Kryl..rod. 10 no.3:28 Nr '59.

(MIRA 12:4)

(Airplanes---Models)

SUMIN, I.V.; GUR'YEV, M.V.; TUNITSKIY, N.N.

Average time of formation of fragment ions from n-hexane. *Kin.*  
kat. 5 no.6:961-967 N-D '64. (MIRA 18:3)

1. Nauchno-issledovatel'skiy Fiziko-khimicheskiy institut imeni  
Karpova, Moskva.

CA

Isotopic composition of the elements. N. Tunitskii.  
*Uspekhi Khim.* 6, 822-9 (1937).—A brief discussion.  
 Radioactivity of potassium and rubidium. *Ibid.* 8:30 2.  
 —Review. F. H. Rathmann

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

COMMON ELEMENTS

COMMON VARIABLE NOTES

*P-6*

*A-1*

Dissociation of the degenerate gas. A. B. SEVERNI and N. N. TUMENI (Physical Z. Soviet-union, 1937, 12, 330-333). - The equilibrium theory for the dissociation of a degenerate gas according to Fermi-Dirac and Bose-Einstein statistics is discussed.

R. S. B.

COMMON ELEMENTS  
COMMON VARIABLE MODELS

OPEN MATERIAL INDEX  
ASS-SLA METALLURGICAL LITERATURE CLASSIFICATION  
REGIONAL INDEX



1ST AND 2ND ORDER										3RD AND 4TH ORDER									
PROCESSES AND PROPERTIES INDEX																			
<p><i>24</i></p> <p>Coagulation of polydisperse systems. N. Tuziński.  <i>Exp. Theoret. Phys. (U.S.S.R.)</i> 8, 417 94(1968).            The case for a const. coagulation coeff. is solved by an            extension of Smoluchowski's method. P. H. Mathmann</p> <p><i>2</i></p>																			
ASD-51A METALLURGICAL LITERATURE CLASSIFICATION																			
FROM SYNDICATE										FROM BUREAU									
1ST AND 2ND ORDER										3RD AND 4TH ORDER									
1ST AND 2ND ORDER										3RD AND 4TH ORDER									

1ST AND 2ND GROUPS										3RD AND 4TH GROUPS									
PROCESSES AND PROPERTIES INDEX																			
<p>02</p> <p>2</p> <p>Formation of aerosols during the condensations of super-saturated vapors. I. V. Petryanov and N. N. Tsumitskii. <i>J. Phys. Chem. (U. S. S. R.)</i> 12, 1131-40 (1939).—Sulfuric acid fogs were produced by mixing streams of SO<sub>3</sub> and H<sub>2</sub>O vapors in capillary tubes, and their properties were detd. by means of an electrometric method. Data are given on the size and concn. of the fog droplets obtained as a function of the concn. and rate of flow of the SO<sub>3</sub> and H<sub>2</sub>O streams, the temp., and the size and length of the capillary tubes. For stearic acid fogs produced by diln. with a cold gas, a diam. of about <math>7 \times 10^{-6}</math> cm. is obtained at temps. up to 187°, above 200° the diam. increases. Comparison of weakly charged aerosols. N. Tsumitskii. 1946. 1141-4.—Theoretical. Equations for calcg. the particle charge of aerosols formed in an ion-free atm. are derived. P. H. Rathmann</p> <p>Moscow Physico-Chem. Inst. in Karpov., Sub. 7 aerosols</p>																			
<p>ASS. S. L. A. METALLURGICAL LITERATURE CLASSIFICATION</p>																			
1ST GROUP										2ND GROUP									
1ST GROUP										2ND GROUP									

TUNITSKIY4N8

600

1. TUNITSKIY, N.
2. USSR (600)

"The Coagulation of Weakly (slabo) Charged Aerosols", Zhur. Fiz. Khim, 13, No. 8, 1939. Moscow, Physico-Chemical Institute imeni L. Ya. Karpov, Lab. of Aerosols. Received 27 Feb 1939.

9. ~~Report~~ Report U-1615, 3 Jan 1952.

BYKOV, R.I.; MAL'TSEVA, A.K.; TURANOV, V.A.; GAVRILOV, V.P.

Regularities in the distribution of oil and gas fields in the  
Jurassic sediments of the central part of the Turan Plateau.  
Trudy MINKHIGP no.43:125-134 '63. (MIRA 17:4)

TURBIN, N. V.; MIRONENKO, A. V.

"The genetic blocking of alkaloid biosynthesis in varieties of fodder lupin."  
report submitted for 10th Intl Botanical Cong, Edinburgh, 3-12 Aug 64.

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AS BSSR, Minsk.

COMMON ELEMENTS		PROCESSING AND PROPERTIES INDEX	
<p><i>ca</i></p> <p>Triboelectrical charges on dust particles. N. M. Tunitshii, M. V. Tikhomirov and I. V. Petryanov. <i>J. Tech. Phys.</i> (U. S. S. R.) 10, 1723 d(1940).-- Triboelec. charges on small particles of coal, ash, Al and marble were measured by the method of N. Rozenblyum (C. A.</p> <p>32, 240P). For ash, Al and marble dust the no. of particles bearing a pos. charge is practically equal to the no. bearing a neg. charge. In coal negatively charged particles predominated somewhat. The surface d. of elec. charge in an atm. of normal humidity was 6-15 elementary charges per <math>1^2</math>, where <math>l</math> is the mean distance between the particles.</p> <p>R. Gamow</p>		<p>4</p>	
<p>ASB-55A METALLURGICAL LITERATURE CLASSIFICATION</p>			
<p>STEEL DIVISION</p>		<p>IRON AND STEEL</p>	
<p>GROUP 1</p>		<p>GROUP 2</p>	
<p>1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100</p>		<p>1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100</p>	

COMMON ELEMENTS		PROCESSES AND PROPERTIES INDEX	
<p><i>Ca</i></p> <p>Charging of dust particles in an electrofilter. N. N. Tushitskii, M. V. Tikhonirov and I. V. Petryanov. <i>J. Tech. Phys.</i> (U. S. S. R.) 10, 1727-37(1940).—The theory of elec. charging of particles proposed by Pauthenier (C. A. 29, 559) was applied to dust particles with triboelec. charges in the case when the accommodation coeff. of ions differs from unity. The expression for the time necessary for full recharging of dust, and for the lowest charge values of these particles which passed the electrofilter were obtained. The discharge process of marble-dust particles in an elec. field is in accord with the theory of Pauthenier. Dry coal particles lose their charge more slowly but the charging of marble and coal particles is very similar. k. Gamow</p>		<p>4</p>	
ASB-51A METALLURGICAL LITERATURE CLASSIFICATION			
MATERIALS INDEX		PROCESS INDEX	
<p>GROUPS</p> <p>1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100</p>		<p>GROUPS</p> <p>1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100</p>	

1ST AND 2ND SECTORS										3RD AND 4TH SECTORS									
PROCEDURES AND PROPERTIES INDEX																			
<p>CA</p>										<p>2</p>									
<p>Particle discharges of an aerosol in a bipolar atmosphere.  N. N. Tunkikh, V. Zarinikh and I. V. Petryanov. <i>Acta  Physicochimica (U.S.S.R. Ser. 12, 527-53 (1946) (in German).—</i>  The particle discharges and that of the whole of a dil.  aerosol in a bipolar-ion atm. proceed according to <math>Z_+ =</math>  <math>Z_-</math>, where <math>Z_+</math> and <math>Z_-</math> are the charges, at 0° and 1° and <math>\rho</math>  is proportional to the ion concn. and independent of the  radius. The Einstein-Fischer equations for the discharge  and its fluctuations are derived. The exptl. data shown  in 7 tables and 4 figs. on oleic acid fogs with <math>r = 0.15 -</math>  <math>0.18 \mu</math> and charged by a Ra prepn., agree well with the  values calcd. from the derived equations.  P. H. Rathmann</p>																			
<p>2</p>																			
<p>ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION</p>																			
<p>ESOM SIMULIV</p>										<p>ESOM BOMIV</p>									
<p>ESOM SIMULIV</p>										<p>ESOM BOMIV</p>									



TUNITSKIY, N.; ZARINSKIY, V.; PELINSKIY, I.

Physico-Chemical Institute imeni L. Ya. Karpov, (-1940-).

"The Charging of Aerosols in a Bipole-Ionized Atmosphere."

Zhur. Fiz. Khim., Vol. 14, No. 4, 1940.

36. 114.

Condensation of superaturated vapours. N. Tunitski. (*J. Phys. Chem. Russ.*, 1941, 25, 1061—1071).—The rate of condensation of  $H_2O$  vapour at adiabatic expansion is calc. The results agree with the experimental data by Wilson (cf. A., 1898, ii, 372) but disagree with those by Harns (*Ann. Physik*, 1907, 24, 226) and Andrén (A., 1917, ii, 192). J. J. D.

*Dr. Ibo* TUNITSKIY, N N.

*41 3 solutions, experiments  
mistaken*

**Influence of van der Waals forces on coagulation of aerosols.**  
M. V. Tikhonov, N. N. Tunitskiy, and J. B. Petrijanov (*Izv. Akad. Nauk SSSR, 1979, 17, 188-190*). An equation is derived which shows that the rate of coagulation of an aerosol is determined uniquely by the mean value of  $R$ ,  $1/R$ , and  $1/R^2$ , where  $R$  is the radius of the particles. On taking into account the interaction between the particles, i.e., the van der Waals forces, the coagulation const. is increased by a factor  $\gamma$ , which is independent of  $R$  but is determined by the const. in the van der Waals forces equation. Experiments with mists of mineral oil, tritoly phosphate, and  $H_2SO_4$  give  $\gamma$  values of 1.25, 1.32, and 1.30, respectively. J. E. H.

"APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001757510003-2

SECRET

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001757510003-2"

1ST AND 2ND ORDERS										3RD AND 4TH ORDERS									
PROCESSES AND PROPERTIES INDEX																			
<p><i>CD</i></p> <p>Diffusion processes in the condition of natural turbulence. N. N. Tunitakii (Karpov Inst. Phys. Chem., Moscow). <i>J. Phys. Chem. (U.S.S.R.)</i> 20, 1137-41 (1946) (in Russian).—Theoretical. The natural turbulence of the atm. can significantly accelerate the coalescence of aerosols, the rate of evapn. of drops, and the heat exchange between drops and the atm. J. J. B.</p>																			
ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION										E-277.576.7.2.2.2.2									
FROM SYMBOLISM										FROM SYMBOLS									
SYMBOLS										SYMBOLS									

Dynamic theory of adsorption and chromatography. I.  
Broadening of chromatographic bands and adsorption fronts.  
N. N. Tunkikh and B. P. Chernova (Karpov Phys. Chem.  
Inst., Moscow), *Dokl. Akad. Nauk*, 24, 1880-81 (1950);  
cf. *Doklady Akad. Nauk* 57, 8, 471 (1947). --  
Chromatographic processes are treated statistically. A  
sharp adsorption band is broadened because of the fluctua-  
tion of the lifetime of the particles in the adsorbed state.  
Two relations are then derived:  $(\sigma - \delta)^2 = 2Kt$  (1) and  
 $(\sigma - \delta)^2 = 2Dx$ , which are valid both when diffusion in the  
liquid or the adsorption is rate-deterg. (1) and when the  
limiting step is the diffusion through the adsorbent particles  
(2). For 1,  $K = v^2/\Delta t$ , where  $v$  is the velocity of the liquid  
relative to a 1-cm. section of the column,  $\delta = \delta_1(1 - \alpha)$ ,  
 $\delta_1 = \alpha(1 - \alpha)$  with  $\alpha$  = relative vol. occupied by the liquid,  
 $\delta_1$  = adsorption coeff.,  $\delta_1$  = kinetic coeff. A linear ad-  
sorption isotherm is assumed throughout. For 2,  $c$  =  
adsorption isotherm is assumed throughout. For 2,  $c$  =  
 $\sigma^2/3l^2D$ , where  $D$  is the diffusion coeff. inside the adsorbent  
and  $l$  is the thickness of the adsorbent particles. Also for  
2,  $H = \sigma^2/3l^2D$ . By taking into account longitudinal dif-  
fusion and hydrodynamic factors, 2 additional terms are obtained  
in the expression for  $K$ , and (1) becomes:  $(\Delta\sigma)^2 = [(v^2/\Delta t) +$   
 $+ \alpha D_1 + (h/\Delta t)]$ , where  $D_1$  is the diffusion coeff. in the flowing  
liquid,  $\alpha$  = const.,  $h$  = f (Reynolds no.) = const. if the  
flow is laminar. Exptl. data obtained with a cation ex-  
changer confirm the theory. First, the column is washed  
with 20 cc. of a 0.1 N soln. (40% NaCl and 60% HCl).

Then the band is displaced with a 0.1 N HCl soln. poured at a  
given rate, and samples (2 cc.) are taken and analyzed (chloride  
of NaCl). The above formulae give  $(\Delta V)^2/V = C_0(2)$   
But  $(\Delta V)^2/V = [(\Delta N)^2/N] \omega$  with  $N = \Sigma N_i/\Sigma C_i$  and  
 $(\Delta N)^2 = [\Sigma(N - N_i)^2 C_i]/\Sigma C_i$ .  $N$  = no. of sample,  $\omega$  =  
its vol.,  $C_i$  = its NaCl concn. A straight line is obtained  
in a diagram  $(\Delta V)^2/V$  against  $\sigma$  in the range 0.3 <  $\sigma$  < 2.1  
cc./min., as predicted by (2). The theory also predicts  
that the distribution function  $H(x, t)$  will be Gaussian at  
least for sufficiently large values of  $t$ . This is verified by  
plotting  $[\log(100/c)]^{1/2}$  against  $\Delta N = N - N_i$ ; a straight line  
is obtained, as expected for a Gaussian distribution.  
Michel Boudart

TUNITSKIY, N.N.

1. Theory of washing chromatographic strips. N. N.

Tunitskii and I. M. Shenderovich (I. Ya. Karpov Phys.-Chem. Inst.). *Doklady Akad. Nauk S.S.S.R.* 81, 649-50 (1951).—The basic parameters of a chromatographic strip (center of gravity, av. width, and the nature of asymmetry) have been detd. by statistical methods. J. Rovtar Leach

THURTSKY, N.M.



KOLOTYRKIN, V.M.; TIKHOMIROV, M.V.; TUNITSKIY, N.N.; SEMENOV, N.N., akademik.

Mass spectrum of methane at increased pressure. Dokl.AN SSSR 92 no.6:1193-  
1195 0 '53. (MLRA 6:10)

1. Akademiya nauk SSSR (for Semenov). 2. Fiziko-khimicheskiy institut im.  
L.Ya.Karpova (for Kolotyrkin, Tikhomirov and Tunitskiy).  
(Methane) (Spectrum analysis)

Tunitskiy, N.N.

USSR/ Physics - Physical chemistry

Card 1/1 : Pub. 147 - 18/22

Authors : Tunitskiy, N. N.; Chernova, E. P.; and Andreyev, V. I.

Title : On the theory of the dynamics of sorption and chromatography. Part 3.-  
The dynamics of ion-exchange sorption during intradiffusion kinetics.

Periodical : Zhur. fiz. khim. 28/11, 2006-2020, November 1954

Abstract : The dynamics of ion-exchange sorption was investigated to determine the behavior of the sorbent layer in a flow containing the sorption substance. The conditions favorable for parallel transfer for ion-exchange sorption are indicated. The term expressing the time loss of the protective effect is introduced. The role of internal diffusion and hydrodynamic factors in the sorption dynamics, is explained. The experimental data obtained for calcium sorption with cationite in H-form, hydrogen sorption with cationite in Ca-form and sorption of radioactive calcium with cationite in Ca-form were found to be in excellent conformity with the theoretical data. Twelve USSR references (1929-1953). Tables; graphs.

Institution : The L. Ya. Karpov Physico-Chemical Institute, Moscow

Submitted : March 26, 1954

TUNITSKIY, N.N.

USSR

537 56.3

4520. Spectrum of methane at elevated pressures. M. K. KUDRYAVTSEV, M. V. LUKHOMIROV, and N. N. TUNITSKIY. *Izv. Akad. Nauk SSSR*, 92, No. 6, 1193-5 (1954) in Russian.

Presents the results of studies on the pressure-dependence of the "triangular" peaks 11<sup>+</sup> and 10<sup>+</sup> occurring in CH<sub>4</sub> as a result of processes CH<sub>4</sub><sup>+</sup> → C<sup>+</sup> + H and CH<sub>4</sub><sup>+</sup> → C<sup>+</sup> + 2H. The dependence of the intensity (*I*) of the primary peaks on pressure (*p*) can be represented by the formula  $I = (4p - Bp^2) \exp(-\kappa p)$ , where 4 is the probability of the primary peak being formed by a collision with an electron, *Bp*<sup>2</sup> is the term allowing for the possibility of a change in the number of residual ions, due to collisions in the ionic source, and  $\kappa$  is the coefficient of scattering. If the coefficients of scattering are equal for the primary and secondary ions, then the intensity of the peaks is proportional to  $I^* = I - Bp^2$ . For *B* = 0 and small values of  $\kappa p$ , the  $I^* \sim p$  law is obtained (C. 1.11) or (C. 4.11). Experiments conducted with a 60 mass spectrometer show that, for the peaks 11<sup>+</sup> and 10<sup>+</sup>, (1)  $I^* \sim p$  is proportional to  $\kappa I \sim pI^2$ , and (2), at small pressure values, proportional to pressure. The cross-section of the reaction CH<sub>4</sub><sup>+</sup> + C<sup>+</sup> → H is of the order of  $1 \cdot 10^{-16}$  cm<sup>2</sup>.

BB

244

TUNITSKIY, N. N.

USSR/Chemistry - Physical chemistry

Card 1/1 Pub. 22 - 24/45

Authors : Tunitskiy, N. N.

Title : The layer method of calculating chromatographic processes

Periodical : Dok. AN SSSR 99/4, 577-579, Dec 1, 1954

Abstract : The layer method for the calculation of chromatographic separation processes, based on the assumption that the continuous process in the column can be considered as a series of equilibria in the small layers of a sorbent, is described. Some results obtained from calculations in accordance with the layer method are listed. Seven references: 6-USSR and 1-USA (1947-1953).

Institution : The L. Ya. Karpov Physico-Chemical Institute

Presented by: Academician M. M. Dubinin, July 3, 1954

*Tunitskiy, N. N.*

USSR/Physical Chemistry - Photochemistry. Radiation Chemistry. Theory of the  
Photographic Process, B-10

Abst Journal: Referat Zhur - Khimiya, No 1, 1957, 470

Author: Tunitskiy, N. N., Kupriyanov, S. Ye., and Tikhomirov, M. V.

Institution: Academy of Sciences USSR

Title: Effect of Electrons of Different Energies on the Ionization and Dis-  
sociation of Some Hydrocarbon Halides

Original

Periodical: Sb. rabot po radiatsionnoy khimii, Moscow, Publishing House of the  
Academy of Sciences USSR, 1955, 223-240

Abstract: The ionization and dissociation of halogen derivatives of hydrocar-  
bons has been carried out with a type MS-1 mass spectrometer, sup-  
plemented as follows: (1) automatic scanning of the mass spectrum,  
(2) automatic recording of the mass spectrum, and (3) introduction  
of the sample into the ion source. The mass spectra of  $\text{CH}_4$ ,  $\text{CH}_3\text{Cl}$ ,  
 $\text{CH}_2\text{Cl}_2$ ,  $\text{CHCl}_3$ ,  $\text{CCl}_4$ ,  $\text{CH}_3\text{Br}$ , and  $\text{CH}_3\text{I}$  have been recorded with electron  
energies of 100 ev. It is shown that as the number of halide atoms

Card 1/2

USSR/Physical Chemistry - Photochemistry. Radiation Chemistry. Theory of the  
Photographic Process, B-10

Abst Journal: Referat Zhur - Khimiya, No 1, 1957, 470

Abstract: In the molecule increases, the maximum intensity of the ionic current shifts from the region of molecular ions to that of ions formed by the splitting off of an atom, i.e., dissociation begins to overshadow ionization. Anomalous ions and some secondary processes which occur during ionization and dissociation were investigated. The formation of  $H_2X^+$  was observed in the mass spectra of some halogen derivatives of methane. The dependence of the ionization and dissociation of the molecules  $CH_3I$ ,  $C_2H_2Cl_4$ ,  $CH_3Br$ , and  $CH_3Cl$  on the electron energy (up to 1,100 eV) was investigated. It is shown that as the energy of the ionizing electrons increases, the mass spectra contained fewer fragment ions.

Card 2/2

KUPRIYANOV, S.Ye.; DZHAGATSPANYAN, R.V.; TIKHOMIROV, M.V.; TUNITSKIY, E.N.

Mass-spectrometric method of analyzing chlorine derivatives of  
methane. Zav.lab.21 no.10:1182-1188 '55. (MLRA 9:1)

1.Nauchno-issledovatel'skiy fiziko-khimicheskiy institut imeni  
L.Ya.Karpova.

(Mass spectrometry) (Methane)

TUNITSKIY, N. N.

USSR/Physics - Physical chemistry

Card 1/2      Pub. 22 - 32/51

Authors :      Tikhomirov, M. V.; Kolotyrkin, V. M.; and Tunitskiy, N. N.

Title :      About the dissociation of primary ions in a mass-spectrometer

Periodical :      Dok. AN SSSR 101/5, 903-905, Apr 11, 1955

Abstract :      The relation between the intensity of "fractional" n-butane peaks and pressure was investigated to explain the mechanism of primary ion dissociation at greater pressures. It is pointed out that the dissociation at greater pressures. It is pointed out that the dissociation during collision, as in the case of spontaneous decomposition, may depend upon the ion excitation and that the excitation varies depending upon the energy of the ionizing electrons. It was found that the relative intensity of the "fractional" peaks increases with the electron

Institution :      The A. A. Zhdanov State University, Leningrad  
Presented by:      Academician A. N. Terenin, November 14, 1954



Card 2/2 : Pub. 22 - 32/51

Periodical : Dok. AN SSSR 101/5, 903-905, Apr 11, 1955

Abstract : energy, this is due to the fact that the spontaneous decomposition of the ions and their decomposition during collisions depend in various degrees upon the electron energy. Eight references: 3 German, 2 USSR, 2 USA and 1 English (1939-1953). Graphs.

TUNITSKIY, N. N.  
USSR/ Physics - Physical chemistry

Card 1/1 Pub. 22 - 28/47

Authors : Tunitskiy, N. N.; Smirnova, R. M.; and Tikhomirov, M. V.

Title : About "broken" peaks in the mass spectrum of hydrogen

Periodical : Dok. AN SSSR 101/6, 1083 - 1084, Apr. 21, 1955

Abstract : A 60° mass spectrometer of low resolving power and magnetic sweep of the mass spectrum was employed in determining the relation between the cross sections of  $H_2^+$  ion dissociation and the energies of the ionizing electron beam. It is shown that the dissociation process results in the formation of a blurred peak (band) in the mass spectrum of hydrogen at an apparent mass of 1/2. It was established that the (1/2) peak, corresponding to dissociation protons, has a width approximately 5 times greater than the basic peak. The relation between dissociation cross section and ion energy is explained. Five references: 4 USSR and 1 German (1939-1954). Graphs.

Institution : The L. Ya. Karpov Sc. Res. Phys. Chem. Inst.

Presented by: Academician V. N. Kondratyev, November 12, 1954

DZHAGATSPANYAN, R.V.; TUNITSKIY, N.N.

~~SECRET~~  
Experimental study of the dilution of chromatographic bands.  
Dokl.AN SSSR 105 no.6:1282-1284 D '55. (MIRA 9:4)

L.Fiziko-khimicheskiy institut imeni L.Ya.Karpeva. Predstav-  
lene akademikom V.A.Karginym.  
(Chromatographic analysis)

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**CIA-RDP86-00513R001757510003-2"**

TUNITSKIY, N. N.

111

TUNITSKIY, N. N. (Prof.)

"The Molecule- and Ionic Dissociation in the Mass Spectrometer."

report presented at Scientific Conference at the Inst. for Physical Chemistry  
imeni L. Ya. Karpov, Acad. Sci. USSR, Nov 1957.

SOV/137-58-7-14194

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 7, p 33 (USSR)

AUTHOR: Tunitskiy, N. N.

TITLE: Theory of Ion-exchange Processes (Teoriya ionoobmennyykh protsessov)

PERIODICAL: V sb.: Materialy Soveshchaniya po primeneniyu ionnogo obmena v tsvetn. metallurgii. Moscow, 1957, pp 4-16

ABSTRACT: Bibliographic entry

1. Ion exchange--Theory

Card 1/1

SOV/137-58-9-18449

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 9, p 38 (USSR)

AUTHORS: Cherneva, Ye. P., Tunitskiy, N. N.

TITLE: Diffusion Through Ion-exchange Membranes (Diffuziya cherez ionnoobmennyye membrany)

PERIODICAL: V sb. Materialy Soveshchaniya po primeneniyu ionnogo obmena v tsvetn. metallurgii, Moscow, 1957, pp 43-47

ABSTRACT: The ranges of technical application of ion-exchange membranes are traced. The rates of permeation of cations through a cationite membrane on both sides of which different solutions were present initially were studied.

G. S.

1. Ion exchange--Theory

Card 1/1



TUNITSKIY, N.N.; TIKHOMIROV, M.V.; KUPRIYANOV, S.Ye.; KOLOTYRKIN, V.M.;  
GUR'YEV, M.V.; POTAPOV, V.K.

Studies in the field of mass spectrometry. Probl.fiz.khim.  
no.1:122-128 '58. (MIRA 15:11)

1. Laboratoriya adsorbtsionnykh protsessov Nauchno-  
issledovatel'skogo fiziko-khimicheskogo instituta im.  
Karpova.

(Mass spectrometry)

TUNITSKIY, N.N.

21(8)

p 3

PHASE I BOOK EXPLOITATION

SOV/1140

1. Vsesoyuznoye soveshchaniye po radiatsionnoy khimii. 1st, Moscow, 1957.

Trudy (Transactions of the First Conference on Radiation Chemistry)  
Moscow, Izd-vo AN SSSR, 1958. 330 p. 4,000 copies printed.

Sponsoring Agencies: Akademiya nauk SSSR. Otdeleniye khimicheskikh nauk, and U.S.S.R. Ministerstvo khimicheskoy promyshlennosti.

Editorial Board: Bakh, N.A. Professor (prep. ed.); Medvedev, S.S., Corresponding Member, Academy of Sciences, USSR; Veselovskiy, V.I., Professor, Dolin, P.I., Doctor of Chemical Sciences; Miller, N.B., Candidate of Chemical Sciences; Tsetlin, B.L., Candidate of Chemical Sciences (Secretary). Eds. of Publishing House: Trifonov, D.N. and Bugayenko, L.T.; Tech. Ed.: Moskvacheva, N.I.

PURPOSE: This book publishes the reports of the First All-Union Conference on Radiation Chemistry in Moscow, March 25 - 30.

COVERAGE: This collection includes fifty-seven reports and follow-up discussions of each sub-group of reports classified as follows:

Card 1/15

Transactions of the First (Cont.)

SOV/1140

- 1) primary functions in radiation-chemical processes,
- 2) radiation chemistry of water solutions (inorganic and organic systems),
- 3) radiation-electrochemical processes,
- 4) the effect of radiation on substances which take part in biochemical processes,
- 5) radiation chemistry of simple organic systems,
- 6) radiation effects on polymers, and
- 7) sources of radiation.

According to the editors, the discussions reveal differences in points of view of Soviet scientists on various problems of radiation chemistry; specifically, the mechanism of the action of radiation on concentrated water solutions, the practical importance of radiation-galvanic phenomena, the mechanism of the action of radiation on polymers, etc. The editors also note that the conference revealed inadequate development in some important areas of radiation chemistry, particularly the theory of initiation of radiolysis, and the action of radiation on solid bodies.

Card 2/15

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SOV/1140

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